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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/629,642	07/29/2003	John W. Evans	79213	6851
22242	7590 11/28/2005	EXAMINER		
FITCH EVEN TABIN AND FLANNERY			OGDEN JR, NECHOLUS	
120 SOUTH L	A SALLE STREET			<del></del>
<b>SUITE 1600</b>			ART UNIT	PAPER NUMBER
CHICAGO, II	L 60603-3406		1751	

DATE MAILED: 11/28/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

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		Application No.	Applicant(s)			
Office Action Summary		10/629,642	EVANS, JOHN W.			
		Examiner	Art Unit			
		Necholus Ogden	1751			
The MAI Period for Reply	ILING DATE of this communication app	pears on the cover sheet with the	e correspondence address			
WHICHEVER I  - Extensions of time after SIX (6) MON'  - If NO period for rep  - Failure to reply with Any reply received	D STATUTORY PERIOD FOR REPL'S LONGER, FROM THE MAILING DOWN THE MAILING THE M	ATE OF THIS COMMUNICATION  36(a). In no event, however, may a reply be will apply and will expire SIX (6) MONTHS for a cause the application to become ABANDO	ON. timely filed om the mailing date of this communication. NED (35 U.S.C. § 133).			
Status						
1)⊠ Respons	ive to communication(s) filed on <u>20 A</u>	pril 200 <u>5</u> .				
<u> </u>	This action is FINAL. 2b) This action is non-final.					
3) Since this	s application is in condition for allowar	nce except for formal matters, p	prosecution as to the merits is			
closed in	accordance with the practice under E	Ex parte Quayle, 1935 C.D. 11,	453 O.G. 213.			
Disposition of Cla	ims					
4)⊠ Claim(s)	1-16 and 18-44 is/are pending in the	application.				
	4a) Of the above claim(s) <u>18-25 and 33-42</u> is/are withdrawn from consideration.					
· ·	is/are allowed.	•				
<u> </u>	 <u>1-16,26-32,43 and 44</u> is/are rejected.					
	is/are objected to.					
8) Claim(s)	are subject to restriction and/o	r election requirement.				
Application Paper	·s	,				
_						
•	ification is objected to by the Examine ing(s) filed on is/are: a)☐ acc		e Evaminer			
·—	may not request that any objection to the	•				
• •	ent drawing sheet(s) including the correct					
· •	or declaration is objected to by the Ex	,	· ·			
Priority under 35 (	U.S.C. § 119					
12) Acknowle	dgment is made of a claim for foreign ☐ Some * c) ☐ None of:	priority under 35 U.S.C. § 119	(a)-(d) or (f).			
· <u>-</u> ·						
	pies of the certified copies of the prior					
	plication from the International Bureaເ		•			
* See the at	tached detailed Office action for a list	of the certified copies not recei	ved.			
Attachment(s)		_				
1) Notice of Referer		4) Interview Summa Paper No(s)/Mail				
	erson's Patent Drawing Review (PTO-948) osure Statement(s) (PTO-1449 or PTO/SB/08) Date	<b>—</b> •	Patert Application (PTO-152)			

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#### Election/Restrictions

1. Applicant's election of claims 1-16, 26-32 and 43-44 in the reply filed on 9-14-05 is acknowledged. Because applicant did not distinctly and specifically point out the supposed errors in the restriction requirement, the election has been treated as an election without traverse (MPEP § 818.03(a)).

# Response to Amendment

1. Claims 1-5, 8-12, 14, 26, 28-30 and 43-44 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 89/09806 to Reny et al.

Reny et al disclose a coolant composition comprising at least 90% by weight of an alkylene glycol and a corrosion inhibiting amount of an inhibitor comprising (a) from 0.02 to 4 parts by weight of an azole, (b) from 0.05 to 3 parts by weight of a molybdate salt and (c) from 0 to 3 parts by weight of phosphoric acid (page 3, lines 1-11). Reny et al further teach that their coolant composition most preferably contains essentially no water (pg. 5, lines 28-34).

Reny et al, however, do not exemplify a coolant composition containing less than 0.5% by weight of water.

It would have been obvious to one of ordinary skill in the art to decrease the amount of water present in example 1 of Reny et al because Reny et al specifically teach that it is preferred that the alkylene glycol is used with essentially no water.

2. Claims 1-16, 26-32 and 43-44 are rejected under 35 U.S.C. 103(a) as being unpatentable over Chemical Abstracts 120:195478 to Cougenhour et al or Chemical

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Abstracts 116:86516 to Dingley or Evans (5,031,579), each in view of Mascioli et al. or Greaney (5,422,026) or Uekusa et al. (5,387,360).

Coughenour et al disclose the use of non-aqueous propylene glycol as an engine coolant (see abstract).

Dingley disclose the use of monoproplyene glycol as the entire engine coolant (see abstract).

Evans '579 discloses a substantially anhydrous coolant comprising propylene glycol (col. 5, lines 50-53) and specifically teaches that said method comprises substantially no water (col. 6, lines 1-3).

Neither Coughenour et al nor Dingley nor Evans '579 disclose the inclusion of molybdate, nitrate or an azole compound.

Mascioli et al disclose an antifreeze composition comprising propylene glycol, sodium molybdate, sodium nitrate, and tolyltriazole (table 3, of example 1).

Greaney disclose an antifreeze composition comprising propylene glycol, sodium molybdate, sodium nitrate and tolyltriazole (table 3 example 1).

Uekusa et al disclose an antifreeze composition comprising propylene glycol, sodium molybdate, sodium nitrate and tolyltriazole (example 8, table 1).

It would have been obvious to one of ordinary skill in the art to add the molybdate, nitrate and tolyltriazole components of either Mascioli et al or Greaney or Uekusa et al to the propylene glycol coolants of Coughenour et al or Dingley or Evans '579 because Mascioli et al or Greaney or Uekusa et al each teach that molybdates, nitrates, and tolyltriazole are effective corrosion inhibitors for propylene glycol coolants, and it

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appears that the propylene glycol coolants of Coughenour et al or Dingley or Evans '579 would benefit from the corrosion inhibition of the additives disclosed by Mascioli et al or Greaney or Uekusa et al, absent a showing to the contrary.

## Response to Arguments

3. Applicant's arguments filed 4-20-05 have been fully considered but they are not persuasive.

Applicant states that because of "fear of water one of ordinary skill in the 1993" would not have understood the cited Evans patent to suggest the use of 100% of propylene glycol and without the use of a buffer." Applicant further argues that one of ordinary skill in the art, based on the '579 patent, would not have used a non-buffered propylene glycol composition which also included corrosion inhibitors additives."

The examiner contends and respectfully disagrees because it is the examiners positions, however, that it appears a buffer is not required to be employed in a composition comprising propylene glycol and either a molybdate, nitrate or azole as shown by Mascioli et al (5,240,631). Note, that the composition of Mascioli et al contains propylene glycol, a molybdate, a nitrate and tolyltriazole, and that no buffer is required (Table 1, column 3, lines 20-30). It is acknowledged that Mascioli et al teach that an alkali metal hydroxide is employed to provide a final pH of 7-10 for concentrate plus water coolant formulation (col. 2, lines 49-51), however the inclusion of the alkali metal hydroxide appears to be limited to the situation where the concentrate is diluted wit a significant portion of water. Therefore, Evans does suggest applicant's claimed invention in view of the Mascioli et al.

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Applicant argues that Coughenour and Dingley articles would not need corrosion inhibitors because as sated by Coughenour "chemical addition should be greatly simplified."

The examiner maintains that it would have been obvious to one of ordinary skill in the art to add the sodium molybdate, sodium nitrate and tolyltriazole corrosion inhibitors of either Mascioli et al or Greaney et al or Uekusa et al to the propylene glycols of Coughenour et al or Dingley because Mascioli et al or Greaney et al or Uekusa et al each teach that sodium molybdate, sodium nitrate and tolyltriazole are effective corrosion inhibitors for propylene glycol coolants and it appears that the propylene glycol coolants of Coughenour et al or Dingley would benefit from the corrosion inhibition of the additives disclosed in Mascioli et al or Greaney et al or Uekusa et al.

Applicant argues that Reny et al do not teach that it is exemplified to use propylene glycol with less than 0.5% water and Reny does not suggest any corrosion inhibitors.

The examiner maintains, however, that Reny et al specifically teach that it is preferable that the alkylene glycol is used with essentially no water (page 5, lines 28-24), wherein it appears that "essentially no water" would encompass amounts of less than 0.5%. Moreover, Reny exemplifies tolyltriazole and sodium molybdate as corrosion inhibitors (examples 1-2).

Applicant states that Mascioli et al or Greaney et al or Uekusa et al teach the use of corrosion inhibitors in aqueous compositions.

The examiner respectfully disagrees and contends that Mascioli et al or Greaney et al or Uekusa et al each teach that the water content is small amounts from 1 to 5% as compared to applicant's less than 0.5%, however, Mascioli et al or Greaney et al or Uekusa et al are relied upon to show that the claimed "propylene glycol soluble additives" are well known in the art.

- 4. The Declaration under 37 CFR 1.132 filed 4-20-05 is insufficient to overcome the rejection of claims 1-16, 26-32 and 43-44 based the last Office action because:
- 5. As stated above in the Response to Arguments section, Declarant states that because of "fear of water one of ordinary skill in the 1993" would not have understood the cited Evans patent to suggest the use of 100% of propylene glycol and without the use of a buffer." Declarant further argues that one of ordinary skill in the art, based on the '579 patent, would not have used a non-buffered propylene glycol composition which also included corrosion inhibitors additives."

The examiner contends and respectfully disagrees because it is the examiners positions, however, that it appears a buffer is not required to be employed in a composition comprising propylene glycol and either a molybdate, nitrate or azole as shown by Mascioli et al (5,240,631). Note, that the composition of Mascioli et al contains propylene glycol, a molybdate, a nitrate and tolyltriazole, and that no buffer is required (Table 1, column 3, lines 20-30). It is acknowledged that Mascioli et al teach that an alkali metal hydroxide is employed to provide a final pH of 7-10 for concentrate plus water coolant formulation (col. 2, lines 49-51), however the inclusion of the alkali metal hydroxide appears to be limited to the situation where the concentrate is diluted

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wit a significant portion of water. Therefore, Evans does suggest applicant's claimed invention in view of the Mascioli et al.

Declarant argues that Coughenour and Dingley articles would not need corrosion inhibitors because as sated by Coughenour "chemical addition should be greatly simplified."

The examiner maintains that it would have been obvious to one of ordinary skill in the art to add the sodium molybdate, sodium nitrate and tolyltriazole corrosion inhibitors of either Mascioli et al or Greaney et al or Uekusa et al to the propylene glycols of Coughenour et al or Dingley because Mascioli et al or Greaney et al or Uekusa et al each teach that sodium molybdate, sodium nitrate and tolyltriazole are effective corrosion inhibitors for propylene glycol coolants and it appears that the propylene glycol coolants of Coughenour et al or Dingley would benefit from the corrosion inhibition of the additives disclosed in Mascioli et al or Greaney et al or Uekusa et al.

Declarant argues that Reny et al do not teach that it is exemplified to use propylene glycol with less than 0.5% water and Reny does not suggest any corrosion inhibitors.

The examiner maintains, however, that Reny et al specifically teach that it is preferable that the alkylene glycol is used with essentially no water (page 5, lines 28-24), wherein it appears that "essentially no water" would encompass amounts of less than 0.5%.

Moreover, Reny exemplifies tolyltriazole and sodium molybdate as corrosion inhibitors (examples 1-2).

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Declarant argues that it would not have been obvious to include corrosion inhibitors that worked in systems with essentially little or some water to work in systems that had essentially no water.

The examiner disagrees and directs applicant's attention to Uekusa et al that teach that there is a need for corrosion resistance to alcohols present in engine coolant formulations (col. 1, lines 14-21). Therefore, one of ordinary skill in the art would look to Uekusa et al. to suggest that the alcohols such as propylene glycols were causing corrosion and therefore the need for corrosion inhibition would not be solely based upon the addition of water but also the use of propylene glycol based solutions.

With respect to subparagraphs 13-14 of the Declaration, the examiner contends that applicant has not sufficiently demonstrated that additives would not be needed in the compositions of Coughenour et al because they would fall from suspension and congeal. This assertion has not been established by comparing the data, commensurate in scope with the claimed invention against the disclosure of Coughenour et al. Therefore, the rejection is maintained for reasons disclosed herein.

### Conclusion

6. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the

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shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Necholus Ogden whose telephone number is 571-272-1322. The examiner can normally be reached on M-Th.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Yogendra N. Gupta can be reached on 571-272-1316. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Necholus Ogden Primary Examiner Art Unit 1751

No 11-19-05